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# IDENTIFICATION OF CALYSTEGINES IN PLANTS OF FAMILY *ERICACEAE* WITH GAS CHROMATOGRAPHY – MASS SPECTROMETRY METHOD

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In Russia, folk-medicinal plants of the family *Ericaceae* are often used for the treatment of diseases associated with hyperglycemic states. It is known that calystegines, a group of compounds of nortropane alkaloids, have the unique ability of competitive inhibition of glucosidases activity. The objective of this study was to identify calystegines in the chemical composition of plants growing on the islands of Solovetsky Archipelago and belonging to the family *Ericaceae: Vaccinium vitis-idaea* (lingonberry or cowberry); *Vaccinium myrtillus* (bilberry); *Oxycoccus palustris* (cranberry); *Empetrum hermaphroditum* (crowberry, a common name also used for corema); *Arctostaphylos uva-ursi* (bearberry) and *Ledum palustre* (marsh tea, wild rosemary).

Preliminary data for the detection of the chemical composition of the studied plants were obtained by capillary zone electrophoresis and further research was carried out by gas chromatography coupled to tandem mass spectrometry.

Thus, the presence of calystegines in plant samples of family *Ericaceae* has been confirmed in the course of the research. It was found that calystegine B2 presents in all samples tested and is dominant. Calystegine B1 was found in cowberry and cranberry. The presence of calystegine B3 was convincingly confirmed only in lingonberry.

The presence of calystegines in the chemical composition of studied plants is one of the reasons of explaining their use in folk medicine for treating of hyperglycemic states.

Keywords: plants, Ericaceae, calystegines, CE, GC-MS/MS.

#### Introduction

In Russia, folk-medicinal plants of the family *Ericaceae* are often used for the treatment of diseases associated with hyperglycemic states. The family *Ericaceae* includes more than 3000 species belonging to 100 genera. In flora of Russia, there are 54 species of this family, and only four species are officinal herbs, *Arctostaphylos uva-ursi L.* (L.) Spreng., *Vaccinium vitis-idaea L., Vaccinium myrtillus* L. and *Ledum palustre L.* [1].

It is known that calystegines, a group of compounds of nortropane alkaloids, have the unique ability of competitive inhibition of glucosidases activity [2]. The objective of this study was to identify calystegines in the chemical com-

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*Mudretsova Iuliia Viktorovna* – Lecturer, Department of Pharmaceutical and Toxicological Chemistry, Candidate of Pharmaceutical Sciences, e-mail: farmnauka@mail.ru position of plants growing on the islands of Solovetsky Archipelago and belonging to the family *Ericaceae*: *Vaccinium vitis-idaea* (lingonberry or cowberry); *Vaccinium myrtillus* (bilberry); *Oxycoccus palustris* (Cranberry); *Empetrum hermaphroditum* (crowberry, a common name also used for Corema); *Arctostaphylos uva-ursi* (bearberry) and *Ledum palustre* (marsh tea, wild rosemary).

Preliminary data for the detection of the chemical composition of the studied plants were obtained by capillary zone electrophoresis (CE) [3] and further research was carried out by gas chromatography coupled to tandem mass spectrometry (GC-MS/MS).

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#### Experimental

Wild herbal species belonging to the family *Ericaceae* and growing on the islands of Solovetsky Archipelago were collected in July and August 2011. Plant materials were identified at the Department of Pharmacy and Pharmacology of the Northern State Medical University, Arkhangelsk, Russia. In the present study, dried leaves of investigated plants were used. Standard samples of calystegines (A3, B1, B2 and B3) were kindly donated by Dr. N. Asano (BioApply Co., Ltd., Kanazawa, Japan).

All other chemicals and solvents were of analytical grade (Sigma Aldrich). Ultrapure water, obtained by a Millipore Milli-Q system (Bedford, MA, USA), was used for standard and sample preparation. Electrolyte solutions were filtered through a 0,45 mm nylon membrane (Supelco, Belle fonte, PA, USA) before use.

Capillary electrophoresis data were obtained using a CE system «Capel 105» (Lumex, Russia) equipped with variable wavelength UV-detector (wavelength range 190–380 nm) and autosampler. The capillary length was 75 cm (effective length 65 cm), internal diameter 50  $\mu$ m, detection wavelength 191 nm, temperature 50 °C; voltage of 25 kV. The injection by pressure was used (50 mbar × 5 sec).

Before use, the capillary washed with 0,5 M solution of hydrochloric acid during 20 min, then by water for 10 min, after that by 0,5 M solution of sodium hydroxide for 20 min. Then the capillary was washed again with water for 10 min. During analysis, the capillary was successively washed with 0,5 M solution of sodium hydroxide for 2 min, followed by water for 2 min, and then equilibrated with the running buffer for 3 min.

The carrier buffer was prepared by dissolving sodium tetraborate in water to give a concentration of 80 mM and, if necessary, adding sodium hydroxide solutions to give a pH exactly 9,2 [3].

GC-MS data were collected using Agilent 7000 QQQ Triple Quadrupole system (Agilent, USA), equipped with capillary column HP-5MS, 0,32 mm i.d. x 30 m, 0,25  $\mu$ m stationary phase. The following parameters of analysis were used. Split ratio 1 : 20. Injector temperature 170 °C. Carrier gas – helium. Control of gas flow – constant speed. The flow rate through the column – 1 ml/min (27,9 cm/sec). Temperature program of thermostat: 150 °C for 3 min, rise at 10 °C/min up to 240 °C. MS interface temperature 230 °C. Ion source temperature 230 °C. The ionization energy – 70 eV. Detector voltage – 1,8 kV. CID gas: nitrogen, 1ml/min. Quench flow: He, 2 ml/min.

The compounds have been identified by combined gas-chromatographic and mass spectrometric analysis (GC–MS/MS) comparing retention times and fragmentation patterns with standard samples.

The standard solutions of calystegines (1 mg/ml) for CE analysis were prepared in 50% aqueous ethanol. All prepared solutions were filtered and centrifuged at 7000 rpm for 5 minutes. To prepare the model mixture including all standard solutions, standard samples were mixed in the ratio 1 : 1 : 1 : 1.

All dried leaf samples of investigated plants, 10 g of each were homogenized in 100 ml 50% aqueous ethanol and allowed to stand for 24 hours at room temperature.

For CE analysis the extracts were filtered through a 0,25 µm filter before injection.

For GC-MS analysis, ethanol-water extracts in each case was filtered. The filtrates were evaporated under vacuum to give brown syrups in amount of about 1 ml. These syrups were loaded onto glass columns packed with the 10 g of Amberlite IR-120B ( $H^+$  form). Sugars and phenolic compounds were washed from the column with distilled water to give a neutral pH value. The calystegines were eluted with 2 M ammonium hydroxide solution. Ammonium was removed from the resulting solutions by evaporation under vacuum at 40 °C to obtain amount of about 1 ml and lyophilized [5].

For derivatization, the dry plant extracts were dissolved in 500  $\mu$ l of pyridine and allowed to stand for 10 min. After that supernatant (40  $\mu$ l) placed into an Eppendorf centrifuge tube, added 70  $\mu$ l silylating agents (HMDS + TMCS 3 : 1). The tubes were tightly capped and the contents were mixed using Vortex. Then tubes kept at 55 °C for 15 min. Then, the tubes were cooled for 30 min to room temperature and centrifuged at 14800 rpm during 5 min. Supernatant was taken into the GC vial and used for analysis.

#### **Results and discussion**

Since complexation of borate ions with vicinal or alternate hydroxyl groups under alkaline conditions gives compounds which can be detected by UV at low wavelength [3], we used capillary electrophoresis method for identification of calystegines in the samples of extracts of *E. hermaphroditum* and *V. vitis-idaea*.

For this aim, the electropherograms of the model mixture of the standards of calystegines were obtained. Typical electropherogram of calystegines model mixture is presented in Fig.1.

The migration time was 425; 447.7; 467.5 and 624.9 sec, peaks efficiencies were 243587; 146029; 19462; 167034 TP for the standard samples of the calystegines B1; A3; B2 and B3, respectively.

The calystegines in plant extracts were identified by the values of migration time.

The electropherograms of calystegines in plant extracts are presented in Fig. 2.

In the course of preliminary studies, the presence of calystegines A3 and B2 in the chemical composition of *E. hermaphroditum*, and calystegines A3 and B3 in chemical composition *V. vitis-idaea* has been shown.

The confirmation of the presence of identified calystegines of B-group and determination them in plant extracts were done using the method of tandem mass spectrometry in order to provide structural information for the identification.

Analysis of the model mixture of trimethylsilyl (TMS)-derivatized standard samples of calystegines by capillary GC demonstrated that retention time for each of them were 10,66, 10,78 and 11,72 min for TMS of calystegines B3, B1 and B2, respectively, in chosen conditions (Fig. 3).

GC analysis of the investigated TMS derivatives of lyophilized plant extracts in scan mass range detected peaks, approximately corresponding to the retention time of authentic samples of calystegines: calystegine B1 (in extracts from *E. hermaphroditum* and *O. palustris* leafs); calystegine B2 (in all plant extracts) and calystegine B3 (only in extract from *V. vitis-idaea* leafs).

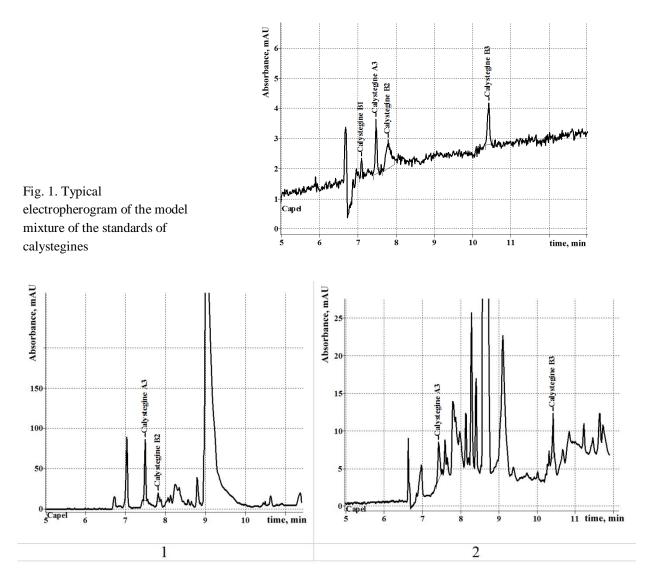


Fig. 2. The electropherogramms of the plant extracts: 1 – E. hermaphroditum; 2 – V. vitis-idaea

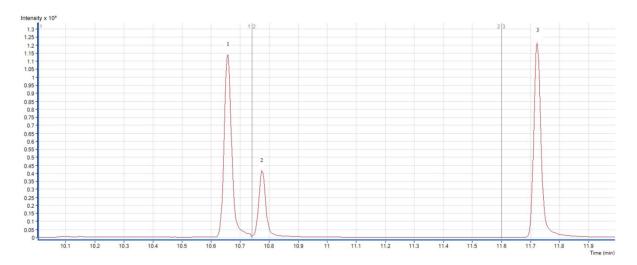


Fig. 3. The chromatogram of the model mixture of TMSi-derivatized standard samples of calystegines: 1 – calystegine B3; 2 – calystegine B1; 3 – calystegine B2

The mass spectra corresponding to these peaks were identified as TMS derivatives of sugars, furanose and pyranose types by base of the Library NIST-08. However, there are a number of peaks in mass spectra, uncharacteristic for TMS derivatives of sugars but typical for TMS derivatives of calystegines, so the analysis selectivity was insufficient because of very complex composition of plant extracts.

We used multiple reaction monitoring (MRM) technique to enhance the selectivity of the analysis for positive identification of calystegines in the tests. By the injection of model mixture of TMS derivatives of calystegine standards the characteristic precursor ions were identified and specific MRM-transitions were selected, the collision energy and the flow of CID gas in the collision cell were optimized to ensure the best signal/noise ratio (SNR) for the target compounds. The obtained conditions are given in Table.

The  $MS^2$  spectra of calystegine B2 prevail in all extracts and its authentic sample was recorded to confirm the presence of calystegines in plant extracts. Recording was carried out in the scan mode product ion monitoring (parent ion 217,5, collision energy 15eV) (Fig. 4).

Presence of calystegines B1 and B3 in the plant extracts was identified by retention times of the peaks in the chromatograms, as well as by compliance ratio of the peak areas of the two MRM-transitions defined for standard samples.

Calystegines	Retention	(Gain) amplifica-	MRM -transitions	Dwell time,	Collisions
	time, min	tion of the signal		msec	energy, eV
B1	10,78	30	244,5→129,1	100	10
			244,5→75,0	100	15
B2	11,72	10	217,5→143,0	100	10
			217,5→73,0	100	20
B3	10,66	10	217,5→143,0	100	10
			217,5→73,0	100	20

The optimal conditions of MRM-analysis of calystegines

CID gas: nitrogen, 1ml/min. Quench flow: He, 2 ml / min

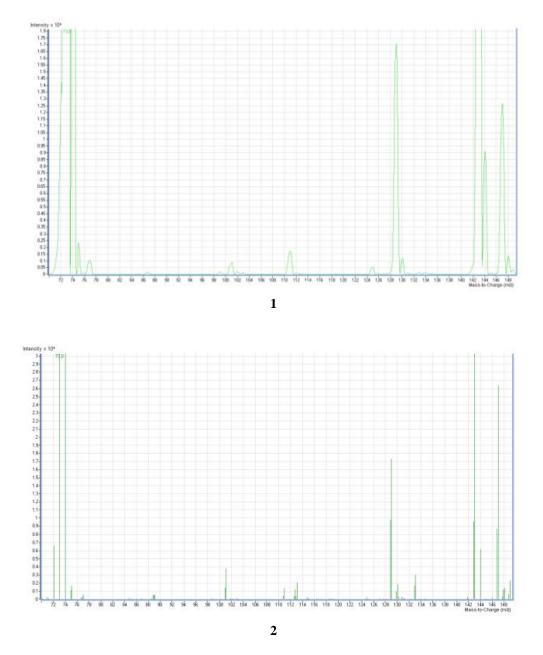


Fig. 4. The spectrum of ion-decay products of calystegine B2: 1 - The spectrum of ion-decay products of calystegine B2; 2 - The typical spectrum of ion-decay products of the compound with retention time 11,72 min in chromatogram of plant extracts

#### Conclusion

The presence of calystegines in plant samples of family *Ericaceae* has been confirmed in the course of the research. It was found that calystegine B2 presents in all samples tested and is dominant. Calystegine B1 was found in lingonberry and cranberry. The presence of calystegine B3 was convincingly confirmed only in lingonberry.

The presence of calystegines in the chemical composition of studied plants is one of the reasons explaining their use in folk medicine for treating of hyperglycemic states.

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